

SAMPLE TEST

English TOLC-F

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BIOLOGY

- 1. The RNA is composed by:
 - A. amino acids
 - B. fatty acids
 - C. nucleotides
 - D. vitamines and carboxylic acids
 - E. carbohydrates

- 2. Fatty acids are made up by:
 - A. hexoses and pentoses
 - B. hydrocarbon chains
 - C. long amino acid chains
 - D. triglycerides
 - E. acetic acid and uric acid

- 3. Bacteria are:
 - A. filamentous only
 - B. algae
 - C. eukaryotes
 - D. unicellular
 - E. multicellular

- 4. Two subsequent cycles of cell division not separated by a phase of DNA synthesis are termed:
 - A. interphase
 - B. anaphase
 - C. meiosis
 - D. mitosis
 - E. double mitosis

- 5. The male reproductive organs in the flower are termed:
 - A. sepals
 - B. stomata
 - C. pistils
 - D. stamens
 - E. petals

- 6. Mitochondria are important:
 - A. in cellular respiration
 - B. in protein synthesis
 - C. in fagocytosis
 - D. in cellular secretion
 - E. in celllular digestion

- 7. If in a cell the ribosome function is selectively blocked, what is immediately stopped?
 - A. Translation
 - B. Transcription
 - C. Glycolysis
 - D. DNA replication
 - E. Respiration

- Consider a phospholipid made up by a head and two tails.
 Which of the following statements is not correct?
 - A. The head contains a phosphate group
 - B. The head contains a molecule of glycerol
 - C. The head is always positively charged
 - D. Tails are hydrophobic
 - E. The head is hydrophilic

- 9. The advantage of sexual reproduction:
 - A. is due to the fact that it has allowed the diffusion of plants on dry lands
 - B. is due to the fact that it happens more frequently than asexual reproduction
 - C. does not exist
 - D. is due to the fact that it increases the genetic variability of organisms
 - E. is less than that given by asexual reproduction

- 10. The process by which an inheritable change is made in a gene is termed:
 - A. splicing
 - B. mutation
 - C. transduction
 - D. alternative splicing
 - E. nick translation

- 11. Which is the product of anaerobic glycolysis from a glucose molecule?
 - A. 3 molecules of lactic acid
 - B. 2 molecules of glycerol
 - C. 2 molecules of piruvic acid
 - D. One molecule of citric acid
 - E. 3 molecules of acetyl coenzyme A

- 12. The terminal section of the large intestine is:
 - A. descending colon
 - B. duodenum
 - C. rectum
 - D. cecum
 - E. pylorus

- 13. In the circulatory system there are, in order:
 - A. heart, vessels for distribution, exchange, and collection
 - B. heart, vessels for distribution, collection, and exchange
 - C. heart, vessels for exchange, distribution, and collection
 - D. vessels for distribution, heart, and vessels for collection
 - E. heart, vessels for collection, distribution, and exchange

- 14. Blood reaches the left atrium:
 - A. from aorta
 - B. with low oxygen
 - C. very rich in oxygen
 - D. from the inferior vena cava
 - E. rich in carbon dioxide

- 15. Steroid hormones:
 - A. are produced in the medulla of the adrenal gland and in the thyroid
 - B. are exclusively testosterone and progesterone
 - C. are produced in the pancreas
 - D. are produced exclusively in the gonads
 - E. are produced in the cortex of the adrenal gland, in the gonads and in the placenta

CHEMISTRY

- 16. Two electrons that share the same orbital have
 - A. different speeds
 - B. different principal quantum numbers
 - C. different masses
 - D. different angular quantum numbers
 - E. different spin quantum numbers

- 17. Which of the following solutions has the lowest value of pH?
 - A. HCI 0,01 M
 - B. NaOH 0,1 M
 - C. HCI 0,2 M
 - D. HCI 0,8 M
 - E. HCI 0,1 M

- 18. The ideal gas law is valid for
 - A. gases dissolved in water
 - B. small molecules
 - C. one mole of a gas
 - D. non-interacting point-like molecules
 - E. inert gases

- In which of the following substances bonds are mainly due to electrostatic forces?
 - A. Sodium chloride
 - B. Hydrochloric acid
 - C. Sodium
 - D. Diamond
 - E. Br₂

20. The chemical properties of an atom are determined by

- A. its atomic mass
- B. its atomic number
- C. its mass number
- D. the number of its neutrons
- E. its metallic properties

- 21. The molecular formula of hydrogen sulphide is
 - A. H_2S
 - B. HNO₃
 - $C. \quad H_2SO_3$
 - $\mathsf{D}. \ \mathsf{H}_2\mathsf{SO}_4$
 - $\mathsf{E}. \quad \mathsf{HNO}_2$

- 22. What is meant by the term orbital?
 - A. The area (around the nucleus) where it is most likely to fiind an electron
 - B. The maximum distance from the nucleus of an electron in the highest energy orbit
 - C. A circle around the atom
 - D. The position of an electron
 - E. The orbit of the electron around the nucleus

- 23. The compound KH is called
 - A. kripton hydroxide
 - B. potassium hydrate
 - C. potassium hydride
 - D. potassium hydroxide
 - E. potassium hydronium

- 24. A millimole
 - A. a unit of measurement for mass
 - B. 1000 moles
 - C. is equivalent to 10^3 moles
 - D. is equivalent to 10^{-3} moles
 - E. 0,01 moles

- 25. Which of the following particles are transferred during redox reactions?
 - A. Neutrons
 - B. Electrons
 - C. Protons
 - D. Nucleons
 - E. Neutrinos

- 26. In the semi-reaction $Cu^{2+} + 2e^- \rightarrow Cu$, the copper
 - A. undergoes oxidation
 - B. undergoes reduction, because it gains two electrons
 - C. does not undergo reduction nor oxidation
 - D. loses two electrons
 - E. undergoes both oxidation and reduction at the same time

- 27. What does organic chemistry deal with?
 - A. Carbon compounds
 - B. The processes going on in animal organs
 - C. The relations between chemistry, animals and vegetals
 - D. The relations between chemistry and minerals
 - E. Nitrogen compounds

- 28. Which of the following substances is a strong acid?
 - A. Carbonic acid
 - B. Acetic acid
 - C. Oleic acid
 - D. Hydrochloric acid
 - E. Sodium hydroxide
- 29. Water molecules are
 - A. non-polar
 - B. none of the answers above is correct
 - C. polar
 - D. completely dissociated into ions
 - E. acid
- 30. In alkanes, the chemical bonds between carbon atoms are
 - A. some π -bonds, some other σ -bonds
 - B. all π -bonds
 - C. all *o*-bonds
 - D. not enough information to answer
 - E. dative bonds

MATHEMATICS

- 31. The expression $4 \times 10^{0} + 5 \times 10^{1} + 3 \times 10^{2} + 7 \times 10^{3}$ is equal to
 - A. 4.735
 - B. 7.354
 - C. 4.537
 - D. 4.530
 - E. 7.350

32. For which values of x is the inequality $x^2 > 36$ satisfied?

- A. x > -6
- B. *x* < −6 or *x* > 6
- C. *x* > 6
- D. *x* > 36
- E. −6 < *x* < 6

- 33. The cosine of a 110° angle is
 - A. negative
 - B. greater than 1/2
 - C. greater than the sine of a 110° angle
 - D. positive
 - E. equal to the cosine of a 290° angle

- 34. Can the logarithm in base 10 of a real number be negative?
 - A. Yes, if the number is negative
 - B. No, never
 - C. Yes, if the number is smaller than 1/10
 - D. Yes, if the number is smaller than the base (10)
 - E. Yes, if the number is positive and smaller than 1

- 35. A quadrilateral can be inscribed in a circumference if and only if
 - A. its adjacent angles are complementary
 - B. the sum of its opposite angles is 120°
 - C. its opposite angles are complementary
 - D. its adjacent angles are supplementary
 - E. its opposite angles are supplementary

- 36. A curve has equation $y^2 = x + 24$. Find the coordinates of the points of intersection between this curve and the straight line x = 1.
 - A. (1,5/2) and (1, -5/2)
 - B. (1,5/4) and (1, -5/4)
 - C. (1,25) and (1, 25)
 - D. (1,5) and (1, -5)
 - E. (5,1) and (-5,1)

- 37. The arithmetic mean of the numbers 3, 4, 5, 6, 7 is
 - A. 3
 - B. 1
 - C. 5
 - D. 0
 - E. 2

PHYSICS

- 38. The prefix mega denotes a factor of
 - A. 10⁹
 - B. 10⁻³
 - C. 10⁶
 - D. 10⁻⁶
 - E. 10⁻⁹

- 39. Stevin's law allows us to find
 - A. the bend radius in a uniform circular motion
 - B. the attractive force between two liquid bodies
 - C. the centripetal acceleration in a parabolic motion
 - D. the pressure exerted by a fluid column at a given depth
 - E. the relative distance between two bodies in uniform accelerated motion

4 - Physics

- 40. Newton's third law of motion states that
 - A. if one body exerts a force on a second body, the second body simultaneously exerts a force equal in magnitude and opposite in direction on the first body.
 - B. the vector sum of forces on a body moving at constant velocity is zero
 - C. a body subject to a force gains an acceleration proportional to such force
 - D. weight is a conservative force
 - E. a body either remains at rest or continues to move at a constant velocity, unless acted upon by a force

- 41. 2 kg of water at 80° C are put into a calorimeter that contains1 kg of water at 20° C. After some time, what equilibrium temperature is reached in the calorimeter?
 - A. 30°C
 - B. 33°C
 - C. 50°C
 - D. 20°C
 - E. 60°C

- 42. The voltage across the terminals of a 150 ohm resistor is 15V. Thus the current flowing through the resistor is
 - A. 1 mA
 - B. 0,1 mA
 - C. 0,1 A
 - D. 10 A
 - E. 1A

- 43. The Avogrado number is the number of molecules contained in
 - A. 1 mole
 - B. 1 mm³
 - C. 1 m³
 - D. 1 cm³
 - E. 1 dm³

- 44. Consider a body in uniform rectilinear motion. Then we can state that
 - A. its velocity is constant
 - B. the force applied to the body is constant
 - C. its acceleration is variable
 - D. its acceleration is constant and non-zero
 - E. its speed is zero

LOGIC

45. Find the next number in the following numerical sequence:

22 - 29 - 25 - 32 - 28 - ?

- A. 32
- B. 35
- C. 39
- D. 38
- E. 29

46. Which pair of numbers completes the following figure?



- A. 4 and 4
- B. 5 and 4
- C. 4 and 6
- D. 4 and 3
- E. 5 and 5

47. Complete the following sequence using the English alphabet (i.e. J, K, W, X, Y are included):

V; Q; M; J; E; A; ?

- A. V
- B. I
- C. X
- D. N
- E. P

48. How much has been the total revenue in Italy and Germany (France excluded) in the second semester of the first year?



- A. Around 130 million euros
- B. Around 70 million euros
- C. Around 40 million euros
- D. Around 90 million euros
- E. Around 60 million euros

- 49. Which of the following is a correct deductive reasoning?
 - A. If Anne went to the gym, and if all the other girls in the gym had brown hair, then Anne would also have brown hair.
 - B. Anne is 6 years old. Girls who are older than 6 are brownhaired. Therefore Anne does not have brown hair.
 - C. Anne is 6 years old. All 6–years–old girls are brown– haired. Therefore Anne has brown hair.
 - D. Anne is 6 years old. Girls who are older than 6 are not brown–haired. Therefore Anne has brown hair.
 - E. Anne goes to the gym. The majority of girls who go to the gym are brown-haired. Therefore Anne has brown hair.
- 50. Three chefs can peel one potato sack in 1, 2, and 4 hours respectively. How many hours will it take to peel 21 potato sacks, if they work together?
 - A. 12 hours
 - B. 10 hours
 - C. 7 hours
 - D. 3 hours
 - E. 6 hours

| Question | Correct Answer | |
|----------|-----------------------|--|
| 1 | С | |
| 2 | В | |
| 3 | D | |
| 4 | С | |
| 5 | D | |
| 6 | A | |
| 7 | А | |
| 8 | С | |
| 9 | D | |
| 10 | В | |
| 11 | С | |
| 12 | С | |
| 13 | А | |
| 14 | С | |
| 15 | E | |
| 16 | E | |
| 17 | D | |
| 18 | D | |
| 19 | А | |
| 20 | В | |
| 21 | А | |
| 22 | А | |
| 23 | С | |
| 24 | D | |
| 25 | В | |
| 26 | В | |
| 27 | А | |
| 28 | D | |

| 29 | С |
|----|---|
| 30 | С |
| 31 | В |
| 32 | В |
| 33 | А |
| 34 | E |
| 35 | E |
| 36 | D |
| 37 | С |
| 38 | С |
| 39 | D |
| 40 | А |
| 41 | E |
| 42 | С |
| 43 | А |
| 44 | А |
| 45 | В |
| 46 | A |
| 47 | С |
| 48 | D |
| 49 | С |
| 50 | А |



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